

SODIUM HEXAMETHYL DISILAZANE

Sodium Hexamethyl Disilazane (SHMDS) in THF 40%

- a. 40 wt% Solution in Tetrahydrofuran
- b. Density at 25°C Approx 0.895 gm/ml

1. OTHER NAMES

- a. Sodium hexamethyl disilazide in THF
- b. Sodium bis (trimethylsilyl) Amides in THF
- c. SHMDS in THF

2. CAS NO.

- a. 1070-89-9 for SHMDS
- b. 109-99-9 for THF

3. FORMULA WEIGHT

183.38 gm/mole

4. TECHNICAL SPECIFICATION

- a. Appearance: Colorless to pale yellow liquid
- b. Total alkalinity (%): 40-42
- c. Hydroxide content (%): 1 max
- d. SHMDS content (%): 39-41

5. SOLUBILITY

SHMDS is soluble in aromatic hydrocarbons and ethers. It is also available in toluene.

6. STABILITY

Atmospheric moisture and carbon dioxide reacts with SHMDS to produce sodium hydroxide and sodium carbonate. Hexamethyl disilazane is liberated from these reactions. The solution becomes cloudy and develops colour. SHMDS solution should be stored in a cool place away from heat, sparks and flame.

7. PACKAGING

- a. Sample packing from 100 gms to 500 gms in glass bottle
- b. 180 kgs in 210 lit. steel drum
- c. Any other packing as per customer request

SUPARNA CHEMICALS LTD
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8. SAMPLING INSTRUCTIONS

- a. The product is packed under dry nitrogen with positive pressure of nitrogen inside the drum.
- b. The quality of the product deteriorates very fast if exposed to atmosphere even for a brief period.
- c. While sampling, please ensure that the sample is taken out under dry nitrogen in a preweighed stoppered bottle and analysis is done immediately.
- d. After sampling, close the container securely after putting positive nitrogen pressure in the drum. This is very important so that the product does not deteriorate on storage.

9. SHIPPING INFORMATION

- a. UN-2920, PG 1
- b. Corrosive flammable liquid

10. PRODUCT PROPERTIES

- a. Very high purity
- b. Very strong base
- c. Selective and specific in many organic reactions
- d. Low hydroxide content
- e. Custom packaging available
- f. Any quantities in bulk

11. PRODUCT BENEFITS

- a. High reaction yields
- USED IN :
- b. Base catalysed reactions
 - c. Strong base for deprotonation reactions
 - d. Superbase reaction used with butyllithium
 - e. Base for Wittig reaction
 - f. Kinetic enolate formation

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